Acta Crystallographica Section E

Structure Reports Online

ISSN 1600-5368

Methyl 4-p-tolyl-1H-pyrrole-2-carboxylate

Michael G. Gardiner,* Roderick C. Jones, Sarah Ng and Jason A. Smith

School of Chemistry, University of Tasmania, Private Bag 75, Hobart, Tasmania 7001, Australia

Correspondence e-mail: michael.gardiner@utas.edu.au

Key indicators

Single-crystal X-ray study T = 193 KMean $\sigma(\text{C-C}) = 0.003 \text{ Å}$ Disorder in main residue R factor = 0.044 wR factor = 0.135 Data-to-parameter ratio = 13.2

For details of how these key indicators were automatically derived from the article, see http://iournals.iucr.org/e.

The molecules of the title compound, $C_{13}H_{13}NO_2$, are close to planar [maximum deviation of 0.0753 (15) Å from the least-squares plane defined by all non-H atoms in the molecule]. Molecules form centrosymmetric dimers through $N-H\cdots O$ hydrogen bonding. Molecules further associate through edge-to-face π -stacking between each face of the p-tolyl substituent and *ortho* H atoms of the p-tolyl units of adjacent molecules.

Received 18 December 2006 Accepted 19 December 2006

Comment

Aryl pyrroles are a common structural motif in many natural products and there has been much research towards synthetic methods for their preparation (Banwell *et al.*, 2006). Recently, we exploited phosphine-free conditions for the Suzuki–Miyaura arylation of iodopyrroles (Smith *et al.*, 2006). Here, we report the synthesis, NMR spectroscopic characterization and X-ray crystal structure of the previously unreported 4-aryl derivative methyl 4-*p*-tolyl-1*H*-pyrrole-2-carboxylate, (I).

The molecular structure of (I) is depicted in Fig. 1, confirming the ring position of the p-tolyl substituent. The p-tolyl and carboxylate substituents lie close to the pyrrolyl ring plane, with the dihedral angles between adjacent rings being 4.30 (12) and 4.52 (12)°, respectively. As a result, the molecule of (I) is close to planar, with a maximum deviation of 0.0753 (15) Å for atom N1 from the least-squares plane defined by the C, N, O atoms.

Molecules of (I) form centrosymmetric dimers in the solid state through N-H \cdots O hydrogen bonding, involving the pyrrolyl N-H and carbonyl O atom of the ester functionality, giving rise to ten-membered $C_4N_2O_2H_2$ rings (Fig. 2 and Table 1). The molecular planes of the constituent molecules are somewhat offset, with the CHNO centres of the hydrogen-bonded ring of each molecule lying out of the related partial molecular plane of the other molecule by between 0.581 (5) and 0.618 (5) Å. In addition, π -stacking between the 4-p-tolyl substituents is apparent, such that *ortho* H atoms of the p-tolyl units of two adjacent molecules interact with each face of the 4-p-tolyl substituent. This gives rise to an infinite three-dimensional arrangement of molecules in the solid state.

© 2007 International Union of Crystallography All rights reserved

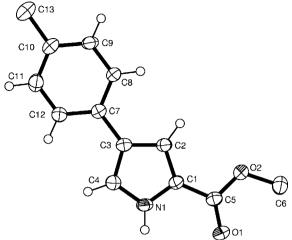


Figure 1

The molecular structure of (I), showing the atom-labelling scheme. Displacement ellipsoids are drawn at the 50% probability level and H atoms are represented by circles of arbitrary size. Disordered methyl H atoms have been omitted for clarity.

Figure 2

A view of the hydrogen-bonding interactions between molecules of (I), shown as dashed lines. Displacement ellipsoids are drawn at the 50% probability level and H atoms are represented by circles of arbitrary size. Disordered methyl H atoms have been omitted for clarity. [Symmetry code: (i) 2 - x, 2 - y, - z.]

Experimental

Compound (I) was prepared according to our literature method (Smith *et al.*, 2006) as colourless crystals in 55% yield. Crystals suitable for X-ray analysis were obtained by recrystallization from diethyl ether–hexane (slow evaporation of the solution in air following filtration of a dilute solution prepared by the addition of diethyl ether to solubilise the suspension of the compound in hexane) (m.p. 448–450 K).

Crystal data

 $\begin{array}{lll} C_{13}H_{13}NO_2 & Z=4 \\ M_r=215.24 & D_x=1.292 \ \mathrm{Mg \ m^{-3}} \\ \mathrm{Monoclinic,} \ P2_1/n & \mathrm{Mo} \ K\alpha \ \mathrm{radiation} \\ a=7.7119 \ (15) \ \mathring{\mathrm{A}} & \mu=0.09 \ \mathrm{mm^{-1}} \\ b=5.4554 \ (11) \ \mathring{\mathrm{A}} & T=193 \ (2) \ \mathrm{K} \\ c=26.385 \ (5) \ \mathring{\mathrm{A}} & \mathrm{Block, \ colourless} \\ \beta=94.61 \ (3)^\circ & 0.55 \times 0.55 \times 0.45 \ \mathrm{mm} \\ V=1106.5 \ (4) \ \mathring{\mathrm{A}}^3 & \end{array}$

Data collection

Enraf-Nonius TurboCAD-4 diffractometer $R_{\rm int} = 0.020$ $\omega/2\theta$ scans $\theta_{\rm max} = 25.0^{\circ}$ Absorption correction: none 3 standard reflections 1975 measured reflections frequency: 60 min 1931 independent reflections intensity decay: none

Refinement

Refinement on F^2 $w = 1/[\sigma^2(F_o^2) + (0.0637P)^2]$ $R[F^2 > 2\sigma(F^2)] = 0.044$ + 0.8352P] where $P = (F_o^2 + 2F_c^2)/3$ S = 1.08 $(\Delta/\sigma)_{\rm max} < 0.001$ $\Delta\rho_{\rm max} = 0.25 {\rm e \ \AA}^{-3}$ $\Delta\rho_{\rm min} = -0.27 {\rm e \ \AA}^{-3}$ Extinction correction: SHELXL97 (Sheldrick, 1997) Extinction coefficient: 0.028 (4)

Table 1 Hydrogen-bond geometry (Å, °).

D $ H$ $\cdot \cdot \cdot A$	$D-\mathrm{H}$	$H \cdot \cdot \cdot A$	$D \cdot \cdot \cdot A$	$D-\mathrm{H}\cdots A$
$N1-H1\cdots O1^{i}$	0.88	2.12	2.955 (2)	159

Symmetry code: (i) -x + 2, -y + 2, -z.

All H atoms were placed in calculated positions and refined using a riding model, with C—H distances in the range 0.88–0.98 Å and with $U_{\rm iso}({\rm H})=1.2U_{\rm eq}({\rm C}),$ except for methyl H atoms, where $U_{\rm iso}({\rm H})=1.5U_{\rm eq}({\rm C}).$ Disordered methyl H atoms were apparent on both the methyl ester and the *p*-tolyl substituent in Fourier difference maps; these were modelled over two sites with occupancies fixed at 0.5.

Data collection: *CAD-4 EXPRESS* (Enraf–Nonius, 1994); cell refinement: *CAD-4 EXPRESS*; data reduction: *XCAD4* (Harms & Wocadlo, 1995); program(s) used to solve structure: *SHELXS97* (Sheldrick, 1997); program(s) used to refine structure: *SHELXL97* (Sheldrick, 1997); molecular graphics: *X-SEED* (Barbour, 2001) and *ORTEP-3* (Farrugia, 1997); software used to prepare material for publication: *WinGX* (Farrugia, 1999).

The authors acknowledge the support of the School of Chemistry in funding this project.

References

Banwell, M. G., Goodwin, T. E., Ng, S., Smith, J. A. & Wong, D. J. (2006). *Eur. J. Org. Chem.* pp. 3043–3060.

Barbour, L. J. (2001). J. Supramol. Chem. 1, 189–191.

Enraf-Nonius (1994). *CAD-4 EXPRESS*. Version 5.1/1.2. Enraf-Nonius, Delft, The Netherlands.

Farrugia, L. J. (1997). J. Appl. Cryst. 30, 565.

Farrugia, L. J. (1999). J. Appl. Cryst. 32, 837-838.

Harms, K. & Wocadlo, S. (1995). XCAD4. University of Marburg, Germany. Sheldrick, G. M. (1997). SHELXS97 and SHELXL97. University of Göttingen, Germany.

Smith, J. A., Ng, S. & White, J. M. (2006). Org. Biomol. Chem. 4, 2477–2482.